Applications of Nanoparticles in Chemistry and Allied Sciences

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_{By} Shazia Syed

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DEDICATED TO

My spiritual father Syed Ilm Ali Shah Jilani and My beloved husband Abdul Faheem Khan

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Preface

Nanoparticles, though incredibly small, are giants in the world of science and technology. "Applications of Nanoparticles in Chemistry and Allied Sciences" is a journey into the fascinating world of these microscopic wonders, illustrating how they're making significant strides in various fields including medicine, electronics, and environmental science. At the heart of this exploration is an understanding of what nanoparticles are — entities so minute, yet so powerful in their ability to change material properties. The book delves into how these particles behave differently from their larger counterparts, revealing a world where size and composition open up a plethora of possibilities. The process of creating these nanoparticles is as fascinating as their applications. The intricate methods of synthesis, the challenge of controlling their size and shape, and the art of tailoring them for specific uses. It's all about the journey through the challenges and triumphs of creating these tiny wonders. But it's not just about what we can create; it's also about understanding the impact of these creations. Nanoparticles' interactions with the environment, their safety, and their potential risks are crucial discussions. It is important to know how these particles behave in different ecosystems and the importance of managing their use responsibly.

As we look to the future, the book also sheds light on the evolving landscape of nanotechnology. It talks about the need for regulations, the ethical considerations, and the endless possibilities that nanoparticles bring to various industries. From advancing medical treatments to revolutionizing electronics and beyond, the potential of nanoparticles seems boundless. "Applications of Nanoparticles in Chemistry and Allied Sciences" is more than just a scientific treatise; it's a narrative that brings the microscopic to the macroscopic level of understanding. It's an invitation to journey through a world unseen, to understand the profound impact of the very small, and to envision a future where nanoparticles play a key role in shaping our world.

Dr. Shazia Syed October 1, 2024

Author Bio



Dr. Shazia Syed is a renowned chemist, deeply immersed in the fascinating field of nanotechnology. She earned her Ph.D. in Chemistry from the University of Karachi, where she also completed her postdoctoral studies. With her teaching and research experience, Dr. Syed's expertise is evident in her book, "Applications of Nanoparticles in Chemistry and Allied Sciences". She skillfully explains how these small particles can make big changes in fields like medicine and environmental science. Her clear, engaging writing style makes the complex subject of nanotechnology accessible and intriguing. This makes her book a great read for anyone curious about nanotechnology. Her dedication to science is also seen in her many publications on chemistry. Dr. Syed is also an active participant in the scientific community. She frequently attends national and international conferences and is a proud member of the American Chemical Society. Her work not only illuminates the path for future research but also inspires a new generation of scientists to explore the boundless possibilities of the microscopic world.

CHAPTER 1 History and background of nanoparticles

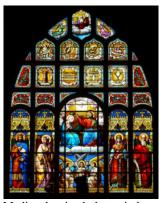
Nanoparticles are nanoscale-sized particles with dimensions typically ranging from 1 to 100 nanometers in at least one dimension. At this scale, materials often exhibit unique and enhanced properties compared to their bulk counterparts. Nanoparticles can be engineered and manipulated for various applications across diverse fields, including medicine, electronics, and environmental science. Nanoparticles are characterized by their minute size, falling within the nanoscale range. This size range imparts distinctive physicochemical properties to nanoparticles (Khan et al., 2019). The high surface area-to-volume ratio of nanoparticles contributes to increased reactivity. Surface interactions become prominent, influencing their behavior and applications (Nel et al., 2006). Quantum phenomena become significant at the nanoscale, affecting the optical and electronic properties of nanoparticles. Nanoparticles can be composed of various materials, including metals, polymers, and ceramics, expanding their utility in different applications (Dreaden et al., 2012). Nanoparticles are extensively used in drug delivery systems, diagnostics, and therapeutic applications due to their size-dependent properties (Farokhzad & Langer, 2009). Nanoparticles play a crucial role in the development of nanoelectronics, quantum dots, and other electronic components (Cui & Lieber, 2001). Nanoparticles contribute to water purification, air filtration, and soil remediation efforts due to their enhanced reactivity. Overall, nanoparticles, defined by their nanoscale dimensions, exhibit unique properties that make them valuable across a spectrum of applications, shaping advancements in science and technology.

Historical background: The history and background of nanoparticles can be traced back to ancient times when artisans unknowingly created materials with nanoscale features. However, the formal recognition and understanding of nanoparticles as a distinct scientific field emerged much later. Artisans and craftsmen unknowingly manipulated materials at the nanoscale while producing stained glass, pottery, and other artifacts. Michael Faraday's work on colloidal gold and silver in the mid-1800s marked an early exploration of particles in the nanometer size range. Early observations and use of nanomaterials can be found in various historical artifacts and materials crafted by ancient civilizations. While the term "nanomaterials" was not used, artisans unintentionally created structures at the nanoscale. The Lycurgus Cup, a Roman glass vessel dating back to the 4th century, exhibits dichroic properties due to the presence of gold and silver nanoparticles (Meeks & Knuuttila, 2009). Currently housed in the British Museum, it is a remarkable example of ancient Roman glasswork. It is renowned for its unusual dichroic (color-changing) properties, where it appears green when lit from the front and red when illuminated from behind. The unique optical properties of the Lycurgus Cup are attributed to the presence of gold and silver nanoparticles dispersed in the glass matrix. High-resolution techniques, including transmission electron microscopy (TEM), have been employed to study thin sections of the Lycurgus Cup, confirming the presence of gold and silver nanoparticles. The nanoparticles in the Lycurgus Cup are believed to be in the range of tens of nanometers in size. These nanoscale structures contribute to the cup's distinctive optical effects. Gold and silver nanoparticles exhibit a phenomenon called surface plasmon resonance (SPR) at the nanoscale. SPR influences the interaction of light with the nanoparticles, contributing to the cup's dichroic behavior. The nanoparticles selectively absorb and scatter specific wavelengths of light, resulting in the observed color changes depending on the angle and direction of illumination. The intentional use of gold and silver nanoparticles in the Lycurgus Cup demonstrates the advanced craftsmanship of ancient Roman glassmakers, showcasing a sophisticated understanding of materials at the nanoscale. The dichroic properties of the cup may have held symbolic or aesthetic significance in the context of ancient Roman art and culture. The Lycurgus Cup serves as an exceptional historical artifact that not only reflects the technological achievements of ancient Roman glassmakers but also provides insights into the early use of nanomaterials for artistic and aesthetic purposes.



Lycurgus Cup in the British Museum

Medieval stained glass windows contain gold nanoparticles, contributing to their vibrant colors (Merrifield, 1849). The presence of nanoscale gold in medieval stained glass windows, created between the 11th and 16th centuries, has been a subject of scientific investigation. While artisans of that era were not aware of nanoscale phenomena, modern analytical techniques have revealed the unintentional incorporation of gold nanoparticles in these artworks. Researchers have utilized transmission electron microscopy to analyze thin sections of medieval stained glass, confirming the presence of nanoscale gold particles. X-ray absorption spectroscopy has been employed to study the chemical composition of stained glass, providing insights into the form and distribution of gold nanoparticles. Nanoscale gold particles contribute to the vivid and rich colors observed in medieval stained glass windows. Gold nanoparticles exhibit unique optical properties at the nanoscale, influencing the way light interacts with the glass and contributing to the overall visual effects (Palomar et al., 2019; Bayda et al., 2019).



Medieval stained glass windows

The study investigates the use of gold nanoparticles in the stained glass of the Santa Chiara Church in Naples, offering insights into the medieval craftsmanship. The unintentional use of nanoscale gold in medieval stained glass windows showcases the craftsmanship of the artisans of that era and provides a fascinating intersection between historical art and modern scientific analysis. The incorporation of such nanoparticles contributes not only to the aesthetics but also to the unique optical properties of these cultural artifacts. Damascus steel, used for forging high-quality blades, is believed to have carbon nanotubes and nanowires, contributing to its strength and sharpness (Reibold et al., 2006). Damascus steel refers to a type of steel that was produced in the Middle East from around 300 to 1700 AD. It is renowned for its distinctive wavy patterns, exceptional strength, and sharpness. The exact methods for producing Damascus steel were lost over time, adding an air of mystery and fascination to this ancient material. One of the most recognizable features of Damascus steel is its distinctive wavy or mottled patterns on the surface of the metal. This pattern is often referred to as the "Damascus pattern" or "Damascus steel pattern." Damascus steel was known for its exceptional strength and durability. Blades made from this steel were highly prized for their ability to maintain a sharp edge and resist damage. In addition to being hard and durable. Damascus steel was often described as having a good balance of hardness and flexibility, making it suitable for creating strong and resilient blades.

The exact methods used to produce Damascus steel have been lost to history, but it is believed that it involved a combination of forging, folding, and heat-treating different types of iron and steel. The production of Damascus steel is associated with the Middle East, particularly regions in present-day Iran and Syria. The steel was highly sought after for the production of high-quality blades, including swords and knives. Damascus steel is often considered a type of crucible steel, which involves melting and solidifying the metal in a crucible. This process can help remove impurities and contribute to the steel's quality. The original methods for producing Damascus steel were lost, possibly due to changes in the availability of raw materials or the secrecy surrounding the techniques. In modern times, attempts to replicate Damascus steel often involve a process known as pattern-welding. This involves layering different types of steel and manipulating them to create patterns similar to those seen in historical Damascus steel (Reibold et al., 2006; Meyers and Rooksby, 1983). The historical significance and allure of Damascus steel lie not only in its exceptional material properties but also in the mystery surrounding its ancient production techniques. The quest to understand and replicate Damascus steel continues to be a subject of interest and research in both historical and metallurgical contexts.

Carbon nanotubes have been found in ancient Chinese ink, suggesting early use of nanomaterials in ink production (Lian & Cui, 2002). The discovery of carbon nanotubes in Chinese ink from the 3rd century is a fascinating revelation that suggests ancient artisans were unintentionally manipulating materials at the nanoscale. While the term "carbon nanotubes" was coined much later in the 20th century, the structural features observed in ancient Chinese ink have similarities to these modern nanomaterials. Researchers have used modern analytical techniques, including transmission electron microscopy (TEM), to examine ancient Chinese ink samples. The presence of carbon nanotubes in the ink has been identified through their characteristic tubular structure. The discovery implies that ancient Chinese artisans were unknowingly manipulating materials at the nanoscale while producing ink. This sheds light on the historical use of nanotechnology. The unintentional inclusion of

carbon nanotubes in ink may have contributed to the ink's unique properties and could reflect a level of technological sophistication in ancient ink-making (Su et al., 2023; Reibold et al., 2006). While not specific to Chinese ink, this study discusses the discovery of carbon nanotubes in an ancient Damascus sabre, highlighting the presence of nanomaterials in historical artifacts. The accidental use of carbon nanotubes in ancient Chinese ink reflects the historical manipulation of materials at a nanoscale level. This discovery not only highlights the unintentional nanotechnology in ancient cultures but also adds to the ongoing understanding of nanomaterials in the broader context of human history.



Damascus steel

Artisans in ancient cultures inadvertently created nanomaterials in pottery glazes, contributing to unique properties (Pradell et al., 2002; Roqué et al., 2006). The unintentional use of nanotechnology in ancient pottery, particularly during the period of 300-500 AD, is an intriguing aspect of historical craftsmanship. While the artisans of that era did not have an understanding of nanotechnology, modern scientific analysis has revealed the presence of nanoscale features in certain pottery glazes. Researchers use advanced techniques such as transmission electron microscopy (TEM) and other nanoscale characterization methods to analyze ancient pottery samples. The presence of nanoscale features, such as nanoparticles or nanocrystals, is observed in the glazes of ancient pottery. The discovery implies that ancient potters were unintentionally manipulating materials at the nanoscale while producing pottery glazes. This demonstrates an early example of working with nanomaterials, even without a formal understanding. Nanoscale features in the glazes may have contributed to enhanced properties of the pottery, such as improved color, hardness, or other characteristics (Pérez-Monserrat et al., 2023). While not specific to the 300-500 AD period, this paper discusses the production technology of Roman ceramics, providing insights into the composition of ancient pottery (Pradell et al., 2002; Roqué et al., 2006). The accidental use of nanotechnology in ancient pottery highlights the craftsmanship and innovation of ancient cultures. While they may not have understood the nanoscale features they were working with, these artisans were able to create materials with enhanced properties, contributing to the historical significance of their creations. These examples illustrate that nanomaterials have been present in various forms throughout history, utilized in artistic creations and functional artifacts. While ancient civilizations did not have a formal understanding of nanotechnology, their craftsmanship inadvertently involved the manipulation of materials at the nanoscale.

Evolution of Nanoparticle Science: The term "nanotechnology" was first introduced by physicist Richard Feynman in his famous 1959 lecture "There's Plenty of Room at the

Bottom," where he discussed the possibility of manipulating individual atoms and molecules. In 1959, the renowned physicist Richard Feynman delivered a seminal lecture titled "There's Plenty of Room at the Bottom" at the annual meeting of the American Physical Society at Caltech. This lecture is often considered the starting point for the concept of nanotechnology. Feynman's vision was centered around the idea of manipulating individual atoms and molecules to create materials and structures with unprecedented precision. Feynman envisioned a future where scientists and engineers would have the ability to manipulate individual atoms and molecules, allowing for precise control over materials. He discussed the possibility of miniaturizing technology by manipulating atoms. suggesting that there is "plenty of room at the bottom" in terms of the small scale at which materials can be manipulated. Feynman speculated on various applications of nanoscale manipulation, including the ability to store information at incredibly high densities, creating new materials with tailored properties, and even the direct manipulation of biological systems. While discussing the challenges of working at such small scales, Feynman emphasized the opportunities for exploration and discovery in this new frontier. Feynman's lecture is often considered the starting point for the conceptualization of nanotechnology. His visionary ideas laid the foundation for the interdisciplinary field of manipulating matter at the nanoscale.

Feynman's vision inspired generations of scientists and researchers to explore the possibilities of nanotechnology. It paved the way for the development of nanoscience and nanotechnology as thriving fields of study. Quotes from the lecture include: "I want to build a billion tiny factories, models of each other, which are manufacturing simultaneously". The principles of physics, as far as I can see, do not speak against the possibility of maneuvering things atom by atom" and "Nature's imagination is so much greater than man's, she's never going to let us relax!" Feynman's vision in 1959 laid the groundwork for the development of nanotechnology, a field that has since seen remarkable advancements and applications in various scientific, engineering, and medical domains. Norio Taniguchi, a Japanese scientist, also played a significant role in the development of nanotechnology. In 1974, he discussed the precision work being done at the atomic and molecular level. His work and the work of other researchers contributed to the development and popularization of nanotechnology as a field of study.

The invention of the Scanning Tunneling Microscope (STM) by Gerd Binnig and Heinrich Rohrer in 1981 allowed researchers to visualize and manipulate individual atoms, marking a significant breakthrough in nanoscale imaging. The Scanning Tunneling Microscope (STM) was not used for nanoparticle synthesis; instead, it is a powerful tool for imaging surfaces at the atomic and molecular levels. The STM was invented by Gerd Binnig and Heinrich Rohrer in 1981, and they were awarded the Nobel Prize in Physics in 1986 for their work. The STM operates based on the principle of quantum tunneling (Duan et al., 2022). A sharp metal tip is brought very close to a surface, and a voltage is applied between the tip and the surface. Electrons can tunnel through the vacuum between the tip and the surface, and the resulting tunneling current is highly sensitive to the distance between the tip and the surface. By scanning the tip across the surface while maintaining a constant tunneling current, a three-dimensional topographic image of the surface can be generated with atomic-scale resolution. The STM has been crucial in advancing nanotechnology by allowing scientists to visualize and manipulate individual atoms on surfaces. However, it is not used for

nanoparticle synthesis. Nanoparticle synthesis typically involves various chemical and physical methods to create nanoparticles with controlled size, shape, and composition. Techniques such as chemical vapor deposition, sol-gel synthesis, and colloidal methods are commonly employed for nanoparticle synthesis and characterization.

In late 20th century, advancements in nanomaterial synthesis methods, such as the development of chemical vapor deposition and sol-gel techniques, facilitated the controlled production of nanoparticles.

In this era, several methods for nanoparticle synthesis were developed, contributing to the foundation of nanotechnology. Some of the key nanoparticle synthesis methods during that time include:

Chemical Precipitation: it involves the mixing of precursor solutions under controlled conditions to induce the precipitation of nanoparticles. This method is very simple and cost-effective. However, there is limited control over size and shape.

Sol-Gel Method: it involves the transformation of a system from a liquid "sol" into a solid "gel" phase to form nanoparticles. It allows for the synthesis of diverse materials, including ceramics and glasses but the control over size and shape may be challenging.

Microemulsion Method: it utilizes the thermodynamic stability of microemulsions to control the synthesis of nanoparticles. Although it provides control over size and shape, good reproducibility but limitations include requirement of surfactants, which may affect the properties of the resulting nanoparticles.

Hydrothermal Synthesis: it involves the reaction of precursor solutions at elevated temperatures and pressures while allowing for the synthesis of high-quality crystalline nanoparticles. Limitations include requirement of specialized equipment and reaction conditions need careful control.

Chemical Vapor Deposition (CVD): it requires gaseous precursors to deposit nanoparticles onto a substrate and provides control over size and composition, suitable for thin film deposition. However, equipment is so complex and may require high temperatures.

Co-precipitation: it is the simultaneous precipitation of multiple cations to form nanoparticles. This method is relatively simple, allows for the synthesis of composite materials. However, control over size and composition may be challenging.

These methods laid the groundwork for the development of more advanced nanoparticle synthesis techniques in the following years. It's important to note that the field of nanotechnology has seen continuous innovation, and newer methods with enhanced precision and control have been developed since the late 20th century.

The discovery of quantum dots and fullerenes (e.g., C60, also known as Buckminsterfullerene) in 1990s highlighted the unique electronic and optical properties of nanoscale materials. Here's an overview of the synthesis methods for these nanoparticles during that period:

Quantum Dots:

Colloidal Synthesis: involves the formation of quantum dots through the colloidal growth of semiconductor nanocrystals and offers control over size, composition, and optical properties. However, it may require stabilizing ligands, and the surface chemistry is crucial.

Organometallic Synthesis: it utilizes organometallic precursors to synthesize quantum dots, often involving high-temperature reactions. Advantage include precise control over size and composition but it requires careful control of reaction conditions, and the process can be complex.

Electrochemical Synthesis: Electrochemical methods to deposit quantum dots on electrode surfaces. It has controlled growth of nanoparticles which is potentially scalable. Limitations include requirement of specialized equipment, and challenging control over size.

Fullerenes:

Arc Discharge Method: involves passing an electric arc through graphite to produce fullerenes. High-yield production of fullerenes is the main advantage of this method with limitations that it requires specialized equipment and separation of different fullerenes can be challenging.

Laser Ablation: this method is based on the principle of laser vaporization of a graphite target to produce fullerenes. High the purity of fullerenes is very high but equipment is very complex, and control over size distribution can be difficult.

Chemical Vapor Deposition (CVD): gaseous carbon-containing precursors are decomposed to form fullerenes in this method. It has a good control over size and structure. Limitations include the requirement of high temperatures and precise control of reaction conditions.

These methods allowed researchers to synthesize quantum dots with tunable electronic and optical properties, and fullerenes with unique structural and electronic characteristics. The ability to control the size, composition, and properties of these nanoparticles opened up new possibilities for applications in fields such as electronics, optics, and materials science.

Emergence of Nanomedicine: Nanoparticles gained prominence in medicine with applications in drug delivery, imaging, and diagnostics. The emergence of nanomedicine as a field gained significant attention in the late 20th century and continued to develop in the 21st century. Nanomedicine involves the application of nanotechnology for medical diagnosis, imaging, and therapy.

In 1990s, development of liposomal doxorubicin (Doxil), a liposomal formulation of doxorubicin, marked one of the early successes in nanomedicine. Liposomes provided a way to improve drug delivery by prolonging circulation time and enhancing accumulation in tumor tissues (Gabizon et al., 1994). Doxil, also known as liposomal doxorubicin, is a medication used in cancer treatment. It is a liposomal formulation of doxorubicin, a commonly used chemotherapy drug. Liposomal doxorubicin is designed to improve the therapeutic index of doxorubicin by encapsulating it in liposomes, which are small vesicles composed of lipid bilayers. Doxil consists of doxorubicin encapsulated in liposomes composed of a phospholipid bilayer, typically pegylated liposomes. The addition of polyethylene glycol (PEG) helps to prolong the circulation time of liposomes in the bloodstream. Like conventional doxorubicin, liposomal doxorubicin exerts its anticancer effects by intercalating with DNA, inhibiting macromolecular biosynthesis, and inducing apoptosis (programmed cell death). The liposomal formulation alters the pharmacokinetics, biodistribution, and toxicity profile of doxorubicin. Liposomal encapsulation helps reduce the cardiotoxicity associated with conventional doxorubicin. Cardiotoxicity is a significant side effect of doxorubicin that can limit its clinical use. The liposomal formulation allows for a more controlled release of the drug (O'Brien et al., 2004).

Liposomal doxorubicin tends to accumulate preferentially in tumor tissues due to the enhanced permeability and retention (EPR) effect. This effect is a result of leaky vasculature in tumors, allowing liposomes to passively accumulate. Liposomal doxorubicin is used in the treatment of various cancers, including ovarian cancer, breast cancer, and Kaposi's

sarcoma. It may be employed as a part of combination chemotherapy or as a single agent, depending on the specific cancer type and stage. While liposomal doxorubicin is associated with a reduced risk of cardiotoxicity compared to conventional doxorubicin, it may still cause side effects such as myelosuppression (bone marrow suppression), hand-foot syndrome, and mucositis. Monitoring for potential side effects is an essential aspect of patient care. Doxil received approval from the U.S. Food and Drug Administration (FDA) for the treatment of certain cancers. It has been an important addition to the armamentarium of cancer treatments (Gordon et al., 2001). Liposomal doxorubicin represents an example of how nanotechnology has been applied to enhance the therapeutic potential of existing drugs and mitigate their side effects. The use of liposomes allows for more precise drug delivery and has contributed to improving the overall safety and efficacy of doxorubicin in cancer treatment.

In 2000s, FDA approval of Abraxane, a nanoparticle albumin-bound formulation of paclitaxel, represented another milestone in nanomedicine. It demonstrated improved efficacy and reduced side effects compared to traditional paclitaxel formulations (Gradishar et al., 2005). Abraxane is a chemotherapy medication used in the treatment of various cancers. It is a brand name for albumin-bound paclitaxel, where paclitaxel, a commonly used chemotherapy drug, is bound to albumin, a protein found in blood. Abraxane consists of paclitaxel, a taxane-class chemotherapy drug, bound to human albumin. The albumin serves as a carrier for paclitaxel. Paclitaxel works by disrupting the microtubule structures in cancer cells, thereby interfering with cell division and causing cell death. The albuminbound formulation is designed to improve the solubility of paclitaxel and alter its pharmacokinetics. Albumin is a naturally occurring protein in the blood. Using albumin as a carrier for paclitaxel has several advantages, including improved drug delivery, reduced toxicity, and potentially enhanced antitumor efficacy. Paclitaxel is a hydrophobic (waterinsoluble) drug. The albumin-bound formulation allows for better solubility of paclitaxel in aqueous solutions, eliminating the need for the solvent used in traditional paclitaxel formulations (Gradishar et al., 2005).

Abraxane is used for the treatment of various cancers, including metastatic breast cancer, non-small cell lung cancer, and pancreatic cancer. It may be used as a single agent or in combination with other chemotherapy drugs, depending on the specific cancer type and stage. Hypersensitivity reactions are a known side effect of conventional paclitaxel formulations. The albumin-bound formulation of paclitaxel has been associated with a lower incidence of severe hypersensitivity reactions. Abraxane received approval from the U.S. Food and Drug Administration (FDA) for the treatment of various cancers. It has become an important option in the management of certain malignancies (Socinski et al., 2012). Abraxane is an example of a nanotechnology-based drug formulation that improves the delivery and performance of a traditional chemotherapy agent. The use of albumin as a carrier enhances drug solubility and alters the pharmacokinetics, contributing to its clinical effectiveness in cancer treatment.

Iron oxide nanoparticles contrast agents became important in magnetic resonance imaging (MRI) as contrast agents. The ability to functionalize these nanoparticles allowed for targeted imaging of specific tissues or diseases (Weissleder et al., 1989). Iron oxide nanoparticles have been extensively studied and used as contrast agents in various medical

imaging techniques, particularly in magnetic resonance imaging (MRI). These nanoparticles provide enhanced contrast in imaging due to their magnetic properties. Superparamagnetic Iron Oxide Nanoparticles (SPIONs): These nanoparticles exhibit superparamagnetic behavior, meaning they become strongly magnetic in the presence of an external magnetic field but lose their magnetization when the field is removed. Ultrasmall Superparamagnetic Iron Oxide Nanoparticles (USPIOs) is a subgroup of SPIONs with smaller sizes, typically less than 50 nm. Iron oxide nanoparticles alter the relaxation times (T1 and T2) of nearby water protons. This change in relaxation times leads to changes in signal intensity, providing contrast in MRI images. Iron oxide nanoparticles are often used for liver imaging, as they are taken up by Kupffer cells in the liver, leading to signal changes in the MRI.

These nanoparticles can be used to enhance visualization of lymph nodes in various anatomical regions. To improve stability, biocompatibility, and targeted delivery, iron oxide nanoparticles are often coated or functionalized with materials such as dextran, polyethylene glycol (PEG), or other polymers. Iron oxide nanoparticles are generally considered biocompatible. They are often cleared from the body through normal physiological processes, primarily through the liver and spleen. Iron oxide nanoparticles are explored for theranostic purposes, combining diagnostic imaging with therapeutic capabilities. They can be loaded with drugs or used for hyperthermia treatment when exposed to an alternating magnetic field (Bulte & Kraitchman, 2004). Iron oxide nanoparticles have contributed significantly to improving the sensitivity and specificity of MRI. Ongoing research continues to explore novel formulations, coatings, and applications of these contrast agents in both diagnostic imaging and therapeutic interventions.

The 2010s saw continued progress in the development of nanoparticle-based cancer therapies. PEGylation (polyethylene glycol coating) of nanoparticles was explored to improve their circulation time and reduce immune responses. Nanoparticle-based cancer therapies have emerged as a promising approach to improve the delivery of anticancer drugs, enhance therapeutic efficacy, and minimize side effects. Liposomal formulations, such as Doxil, encapsulate chemotherapy drugs, allowing for controlled release, improved pharmacokinetics, and reduced toxicity. Polymeric Nanoparticles: Nanoparticles made from biocompatible polymers can carry drugs and release them at the tumor site, offering sustained drug delivery. Nanoparticles can be engineered with ligands that target specific receptors overexpressed on cancer cells, enhancing the selectivity of drug delivery and minimizing damage to healthy tissues. Enhanced Permeability and Retention (EPR) effect allows nanoparticles to accumulate preferentially in tumor tissues due to leaky blood vessels, improving drug delivery to the target site. Nanoparticles enable the simultaneous delivery of multiple therapeutic agents (chemotherapeutics, targeted drugs, nucleic acids) for combination therapies, enhancing treatment efficacy (Jokerst et al., 2011).

Nanoparticles can incorporate imaging agents (e.g., fluorescent dyes, contrast agents) for real-time monitoring of drug distribution and response to therapy, leading to personalized treatment strategies. Nanoparticles are utilized for the delivery of RNA-based therapeutics, including small interfering RNA (siRNA) and microRNA, as well as gene therapy vectors for targeted genetic interventions. Magnetic nanoparticles can generate heat when exposed to an alternating magnetic field, leading to hyperthermia. This approach is explored for localized heating of tumors to enhance the effects of radiation or certain chemotherapeutic

drugs. Nanoparticles are investigated as carriers for immunomodulatory agents, facilitating the delivery of immune checkpoint inhibitors, vaccines, or other immunotherapeutic agents to enhance the body's immune response against cancer cells. Some nanoparticles, such as gold nanorods or carbon-based nanoparticles, can absorb near-infrared light and convert it into heat, enabling photothermal therapy to selectively destroy cancer cells. The development of biodegradable and biocompatible nanoparticles is crucial to minimize long-term side effects and facilitate the clearance of nanoparticles from the body (Bobo et al., 2016). Nanoparticle-based cancer therapies represent a rapidly evolving field with the potential to transform cancer treatment strategies. Ongoing research continues to refine nanoparticle design, optimize drug delivery systems, and explore innovative therapeutic approaches.

RNA nanotechnology gained attention for its potential applications in medicine, including drug delivery and imaging. This represented a new frontier in the development of nanomedicines. RNA nanotechnology involves the design and engineering of RNA molecules for various applications in medicine, including drug delivery, diagnostics, and therapeutics. RNA can be engineered into specific nanostructures, including nanoparticles and nanocarriers, with defined shapes and sizes for various biomedical applications. RNA molecules can spontaneously self-assemble into intricate structures, providing a versatile platform for the development of nanoscale devices. RNA nanoparticles can serve as carriers for delivering therapeutic RNA molecules, such as small interfering RNA (siRNA), microRNA (miRNA), or messenger RNA (mRNA), to target cells. RNA nanocarriers can be designed to target specific tissues or cells through ligand conjugation, improving the precision of drug delivery. RNA nanoparticles are explored as gene delivery vectors for gene therapy applications. They can deliver therapeutic genes to correct genetic disorders or modulate gene expression for therapeutic purposes. RNA molecules can be engineered with imaging moieties (fluorophores, contrast agents) for diagnostic imaging purposes, enabling the visualization of specific cellular processes (Guo, 2010).

RNA nanotechnology enables the development of theranostic platforms, where RNA-based agents serve dual roles in diagnostics and therapy. RNA nanoparticles, particularly messenger RNA (mRNA), have gained prominence in the development of vaccines, such as the mRNA COVID-19 vaccines. RNA aptamers, which are short, single-stranded RNA molecules, can be designed to bind to specific targets, including proteins or other biomolecules, for diagnostic or therapeutic purposes. RNA is a naturally occurring biomolecule, and RNA nanoparticles can be designed to be biocompatible and biodegradable, minimizing potential toxicity concerns (Jasinski & Haque, 2016). RNA nanotechnology holds great promise for advancing personalized medicine and targeted therapies. Ongoing research continues to explore the design principles of RNA nanotechnology, optimize delivery strategies, and expand the range of applications in medicine. This highlights some of the key developments in the field of nanomedicine. It's important to note that nanomedicine is a rapidly evolving field, and ongoing research continues to explore new nanomaterials and applications for improving medical diagnosis and treatment.

Contemporary Nanotechnology: Nanotechnology has evolved into a highly interdisciplinary field, with contributions from physics, chemistry, materials science,

biology, and engineering. Nanotechnology is revolutionizing medicine, offering new approaches to drug delivery, imaging, and diagnostics. Advances include targeted drug delivery systems, imaging agents, and theranostic platforms for personalized medicine (Peer et al., 2007). Nanomedicine is a multidisciplinary field that combines nanotechnology with medicine, aiming to improve diagnostics, drug delivery, imaging, and overall medical treatments. Here are key aspects of nanomedicine along with some relevant references: Liposomes and other nanocarriers are used for drug delivery to enhance drug stability, bioavailability, and targeted delivery to specific cells or tissues. Nanoparticles are employed for targeted drug delivery in cancer therapy, improving the efficacy and reducing side effects of chemotherapy. Nanoparticles, including gold nanorods and quantum dots, serve as contrast agents for advanced imaging techniques like photoacoustic imaging (Jokerst et al., 2012).

Iron oxide nanoparticles are widely used as contrast agents in MRI, offering enhanced imaging of tissues and organs. Nanomaterials are employed in biosensors, enabling the detection of biomolecules at low concentrations for diagnostics and monitoring. RNA nanoparticles are explored for drug delivery and therapeutic applications, including RNA interference (RNAi) and gene therapy. Nanoparticles are used in vaccine design to improve immune responses and enhance vaccine delivery. Theranostic nanoparticles combine diagnostics and therapeutics, allowing for real-time monitoring and personalized medicine approaches (Jokerst et al., 2011). These references provide insights into the diverse applications of nanomedicine. Ongoing research in the field continues to push the boundaries of nanotechnology for innovative medical solutions.

Nanoelectronics involves the development of electronic components at the nanoscale. Silicon nanowires, quantum dots, and other nanomaterials are explored for applications in faster and more energy-efficient electronic devices (Cui & Lieber, 2001). Nanoelectronics refers to the study and application of nanotechnology in the field of electronics. It involves the use of nanoscale materials, structures, and devices to create electronic components and systems. Nanoelectronics aims to overcome the limitations of conventional electronics by exploiting the unique properties of materials at the nanoscale. Semiconductor Nanomaterials: Nanoscale semiconductors, such as nanowires and nanotubes, exhibit novel electronic properties and are used to build electronic devices at the nanoscale. Quantum dots are semiconductor nanoparticles with quantum confinement effects, allowing for precise control of electronic properties. Quantum dot transistors exploit the quantum confinement effects to control the flow of electrons, enabling smaller and more efficient transistors. Nanowires, particularly those made from materials like silicon or carbon nanotubes, can be used to create transistors with enhanced performance. Molecular electronics involves the use of individual molecules as electronic components. This field explores the electronic properties of organic molecules and their potential for building nanoscale circuits.

As traditional complementary metal-oxide-semiconductor (CMOS) technology faces challenges at smaller scales, nanoelectronics explores alternative approaches such as spintronics, quantum computing, and other post-CMOS technologies. Quantum bits or qubits, which can exist in multiple states simultaneously, are investigated for quantum computing. Nanoscale components play a crucial role in the development of quantum

computers. NEMS devices involve the integration of mechanical elements at the nanoscale. Nanoresonators, for example, can be used in sensors and signal processing.

Nanoelectronics relies on advanced fabrication techniques, including top-down methods like electron beam lithography and bottom-up methods like self-assembly of nanomaterials. Nanoelectronics enables the development of ultra-dense memory devices with high storage capacities. Nanoscale transistors contribute to the creation of faster and more energy-efficient electronic devices (Heath & Ratner, 2003). Nanoelectronics holds the potential to revolutionize the electronic industry by enabling the development of smaller, faster, and more energy-efficient devices. Ongoing research in nanoelectronics explores new materials, device architectures, and fabrication techniques to push the boundaries of electronic technology.

Nanophotonics focuses on manipulating light at the nanoscale. Applications include improved solar cells, high-resolution imaging, and novel optical communication devices (Atwater & Polman, 2010). Nanophotonics is a field of study that focuses on the interaction of light with structures at the nanoscale. It involves the use of nanotechnology to manipulate and control light, leading to a variety of applications in optics, electronics, and communications. Plasmonics involves the study of surface plasmon resonances, where free electrons on a metal's surface interact with light. This is exploited for applications like sensing and imaging. Localized plasmons in nanostructures, such as nanoparticles, enable enhanced light-matter interactions and are used in sensing and imaging. Metamaterials are engineered materials with properties not found in nature. Negative refractive index metamaterials are designed to manipulate light in unconventional ways, enabling novel optical devices.

Photonic crystals are periodic nanostructures that can control the flow of light, leading to applications in optical communication and sensing. Nanoscale waveguides confine and guide light at dimensions smaller than the wavelength, allowing for compact photonic devices. Plasmonic waveguides exploit surface plasmon resonances for guiding and manipulating light at the nanoscale. Nanophotonics enables the creation of efficient single-photon sources, essential for quantum information processing and quantum communication. Nanoscale lasers, including plasmonic and semiconductor nanolasers, have applications in data communication and sensing. Nanophotonics plays a crucial role in label-free optical sensing, allowing for highly sensitive detection of biological and chemical substances. Plasmonic nanostructures enhance Raman scattering signals, enabling highly sensitive molecular detection.

Plasmonic nanostructures are employed in solar cells to enhance light absorption and improve the efficiency of energy conversion (Fan & Joannopoulos, 2002). Nanophotonics has significant implications for advancing technology in areas such as telecommunications, sensing, imaging, and energy harvesting. The ability to control and manipulate light at the nanoscale opens up new possibilities for creating more efficient and compact optical devices. Ongoing research in nanophotonics continues to explore novel materials and design principles for the development of advanced optical technologies. Nanotechnology is applied to develop advanced materials for energy storage, such as lithium-ion batteries. Nanomaterials enhance energy density, charge/discharge rates, and overall performance

(Nitta et al., 2015). Nanomaterials play a crucial role in advancing energy storage technologies, offering improved performance in terms of energy density, charge/discharge rates, and overall efficiency.

Ancient Times

4th Century AD: Use of colloidal gold and silver in glass and ceramics (e.g., Lycurgus Cup)



Middle Ages (~9th-17th Century)

9th Century AD: Damascus steel blades containing carbon nanotubes and nanoparticles, known for their strength and sharpness



1850-1900: Early Scientific Observations

1857: Michael Faraday discovers the formation of colloidal gold nanoparticles, leading to the first known scientific experiment involving nanoparticles.

1870s: Lord Rayleigh explains the scattering of light by small particles, foundational for understanding nanoparticle behavior.



1950-1970: Modern Foundations

1959: Richard Feynman gives the famous talk "There's Plenty of Room at the Bottom," proposing manipulating matter at the atomic level.

1960s-1970s: Development of electron microscopy allows the visualization of nanoparticles.



1980-1990: Birth of Nanotechnology

1981: Scanning Tunneling Microscope (STM) invented by Gerd Binnig and Heinrich Rohrer, enabling the observation of atoms and nanoparticles.

1985: Discovery of Fullerenes (buckyballs) by Harold Kroto, Robert Curl, and Richard Smalley.



1990-2000: Rapid Progress

1991: Discovery of **Carbon Nanotubes** by Sumio lijima, further advancing nanotechnology applications.

1990s: Advancements in nanomaterials, like quantum dots, used in electronics, medical imaging, and solar cells.



2000-Present: Commercial and Medical Applications

2000s: Nanoparticles increasingly used in medicine, electronics, energy, and consumer products.

2010s-Present: Ongoing innovations in **nanomedicine**, including drug delivery systems and cancer treatment using nanoparticles.

Historical background of nanoparticles

Nanomaterials, such as nanostructured anodes (e.g., silicon nanowires) and cathodes (e.g., metal oxides, sulfides), enhance the performance of lithium-ion batteries by addressing issues like volume expansion and improving ion diffusion kinetics. Nanomaterials like carbon nanotubes, graphene, and conductive polymers are used in supercapacitors to

increase surface area and enhance charge storage capacity. Nanostructured sulfur cathodes and various carbon-based materials are employed in lithium-sulfur batteries to overcome challenges related to the dissolution of polysulfides and enhance overall performance. Nanostructured materials, such as transition metal oxides and phosphates, are explored for use in sodium-ion batteries as alternative energy storage systems (Manthiram et al., 2014; Simon & Gogotsi, 2008; Kubota et al., 2015). Nanomaterials are investigated for use in solid-state batteries, offering potential advantages in terms of safety, energy density, and cycle life compared to traditional liquid electrolyte batteries. Nanomaterials, including metal hydrides and complex hydrides, are studied for their potential in hydrogen storage applications (Manthiram, 2017; Yildirim & Günaydın, 2012; Chen et al., 2009; Liu et al., 2017). These references provide insights into the diverse applications of nanomaterials in energy storage technologies. Ongoing research in the field continues to explore new nanomaterials and design strategies to further enhance the performance and sustainability of energy storage devices. Nanocomposites, combining nanoscale materials with traditional materials, offer improved mechanical, thermal, and electrical properties, Graphene-based nanocomposites, for example, have applications in various industries (Chu et al., 2017). Nanocomposites in materials science refer to materials that consist of a combination of two or more phases with at least one phase having dimensions in the nanometer range.

The integration of nanoscale reinforcements into traditional materials can lead to enhanced properties and performance. Polymer Nanocomposites are incorporation of nanoscale fillers, such as nanoparticles or nanoclays, into polymers to improve mechanical, thermal, and barrier properties. While metal matrix nanocomposites have nanoscale reinforcement (e.g., carbon nanotubes, graphene) in metallic matrices for improved strength, stiffness, and conductivity. Ceramic nanocomposites involve integration of nanoscale ceramic particles or fibers in ceramic matrices for enhanced toughness and fracture resistance. High aspect ratio CNTs are used to enhance mechanical strength and electrical conductivity in various nanocomposites. Single or few-layer graphene sheets are incorporated to improve mechanical, electrical, and thermal properties. Layered silicate nanoparticles, such as montmorillonite, are added to polymers to enhance mechanical strength and flame resistance (Sanchez-Soto et al., 2019). Nanoparticles in polymers can provide enhanced stiffness, strength, and toughness. Nanocomposites exhibit improved gas barrier properties, making them suitable for packaging applications.

Lightweighting: Nanocomposites in metal matrices can reduce weight while maintaining strength, making them attractive for aerospace and automotive applications. Incorporating nanoscale reinforcements can enhance electrical conductivity in metal matrix composites. Nanoscale reinforcements improve the fracture toughness and strength of ceramic materials. Enhanced thermal stability is achieved in ceramic matrix nanocomposites for high-temperature applications. Automotive Industry: Nanocomposites are used in automotive parts to reduce weight, enhance fuel efficiency, and improve mechanical properties. Nanocomposites find applications in electronic devices for thermal management and electromagnetic interference shielding. Polymer nanocomposites are employed in biomedical applications for controlled drug delivery and tissue engineering (Mehdikhani et al., 2019). Nanocomposites play a vital role in advancing materials science, offering a wide range of tailored properties for various applications. Ongoing research continues to explore

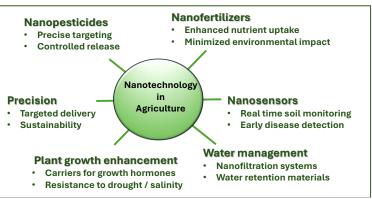
new nanomaterials and optimize processing techniques to unlock the full potential of nanocomposites in different industries.

Nanomaterials are explored for water purification, including the removal of pollutants, heavy metals, and pathogens. Nanotechnology offers efficient and sustainable solutions for water treatment. Nanotechnology has found diverse applications in water treatment, offering innovative solutions to address challenges related to water quality, purification, and remediation (Qu et al., 2013). Nanoparticles, such as silver nanoparticles, titanium dioxide (TiO2), and iron-based nanoparticles, are employed for their antimicrobial and catalytic properties to disinfect water and remove contaminants. Nanocomposites: Composite materials containing nanoscale components, such as carbon nanotubes or graphene oxide, are used to enhance adsorption and filtration processes in water treatment. Nanomaterials like TiO2 and zinc oxide (ZnO) nanoparticles are used in photocatalytic processes, activated by UV light, to disinfect water by degrading organic pollutants and killing microorganisms. Silver nanoparticles exhibit antimicrobial properties and are utilized for disinfection in water treatment. Thin-film composite membranes containing nanomaterials are designed for nanofiltration processes to remove contaminants at the molecular level while maintaining high water flux. Carbon nanotubes are incorporated into membranes to enhance permeability and selectivity for water treatment applications.

Nanoscale adsorbents, such as nanoparticles or nanocomposite materials, are used to remove heavy metals, organic pollutants, and other contaminants from water through adsorption processes. Nanoparticles, including zero-valent iron nanoparticles, are applied in nanoremediation to degrade and remove contaminants, such as chlorinated solvents and heavy metals, from groundwater and soil. Nanotechnology is employed to develop nanoscale sensors capable of detecting and monitoring water quality parameters, such as pH, chemical contaminants, and pathogens (Zhang & Chen, 2016). Nanotechnology offers the potential to revolutionize water treatment processes, providing efficient and cost-effective solutions for improving water quality and ensuring access to clean water. Ongoing research continues to explore new nanomaterials, processes, and technologies to address evolving challenges in water treatment and environmental sustainability. Nanotechnology is applied in agriculture for controlled release of fertilizers, enhanced nutrient uptake, and improved crop protection (DeRosa et al., 2010). Nanotechnology has promising applications in agriculture, aiming to enhance crop yield, improve nutrient utilization, and address challenges related to pest control and environmental sustainability. Some key aspects of nanotechnology in agriculture:

- 1. Nanopesticides: Nanoparticles are used to develop nanopesticides with improved delivery and efficacy, reducing environmental impact compared to traditional pesticides.
- 2. Nano-Fertilizers: Reference: Raliya, R., & Saharan, V. (2018). An overview of commercially available nanofertilizers. Journal of Nanoparticle Research, 20(3), 1-24. Nanofertilizers enhance nutrient availability and uptake by plants, promoting more efficient fertilizer use and reducing environmental nutrient runoff.
- 3. Nanosensors for Precision Agriculture: Nanosensors assist in monitoring soil conditions, nutrient levels, and plant health, contributing to precision agriculture and optimized resource management.

Nanoencapsulation of agrochemicals allows for controlled release, improving the efficiency of nutrient and pesticide delivery to plants. Nanocarriers are explored for targeted delivery of pesticides and nutrients to specific plant tissues, minimizing environmental impact. Nanomaterials are investigated for soil remediation, helping to remove pollutants and improve soil quality (Kah & Hofmann, 2014; Singh et al., 2019; Tripathi et al., 2017; Servin et al., 2015; Khot et al., 2012). Nanotechnology in agriculture has the potential to revolutionize farming practices, contributing to sustainable and efficient crop production. Ongoing research aims to further understand the interactions between nanomaterials and plants, optimize their applications, and address potential environmental concerns.



Nanotechnology in agriculture

Nanotechnology plays a crucial role in the development of quantum computers, leveraging the unique properties of quantum systems for computing applications (Awschalom et al., 2013). Quantum computing is a revolutionary paradigm of computation that utilizes the principles of quantum mechanics to process information. Unlike classical computers that use bits to represent either a 0 or a 1, quantum computers use quantum bits or qubits. Qubits can exist in multiple states simultaneously, thanks to a quantum property called superposition. This property enables quantum computers to perform certain calculations much more efficiently than classical computers for specific tasks. Qubits are the fundamental units of quantum information. Unlike classical bits, which can be in a state of 0 or 1, qubits can exist in a superposition of both states simultaneously. Superposition allows quantum computers to process multiple possibilities at the same time, providing a potential for exponential speedup in certain algorithms. Entanglement is a quantum phenomenon where the states of two or more qubits become correlated and interdependent. Changes to one entangled qubit will instantly affect the other, regardless of the distance between them. Entanglement is a crucial resource for quantum computing algorithms. Quantum gates are the building blocks of quantum circuits, similar to classical logic gates. Quantum gates manipulate qubits to perform specific operations.

Common quantum gates include Hadamard gates, Pauli gates, and CNOT gates. Superposition allows quantum computers to explore multiple solutions to a problem simultaneously. This capability is exploited in algorithms like Shor's algorithm for integer

factorization and Grover's algorithm for database searching. Quantum interference is a phenomenon where different paths to the same outcome interfere constructively or destructively. Quantum algorithms use interference to amplify correct solutions and cancel out incorrect ones. Quantum parallelism arises from superposition, allowing quantum computers to perform parallel computations for certain tasks, providing a potential speedup over classical computers (Arute et al., 2019).

Ongoing Research and Applications: Ongoing research in the field of nanoparticles spans a wide range of applications, from medicine to energy, electronics, and environmental remediation. Recent advancements in nanomedicine include use of nanoparticles in drug delivery for cancer therapy, highlighting the potential of nanocarriers for targeted and controlled drug release (Peer et al., 2007). These carriers can enhance the therapeutic or diagnostic efficacy of payloads by improving their stability, bioavailability, and targeted delivery. Another focus of current research is the use of nanomaterials for scalable energy storage solutions, including batteries and supercapacitors. Ongoing research focuses on optimizing nanomaterial designs and synthesis methods to address the increasing demand for scalable and sustainable energy storage technologies (Choudhury & Archer, 2020).

Silicon nanowires are one-dimensional nanostructures composed of silicon atoms. They exhibit unique electronic, optical, and mechanical properties, making them valuable for various applications, especially in electronics and nanotechnology. Silicon nanowires could be used for high-performance field-effect transistors, demonstrating the potential of nanomaterials in electronics. Semiconductor nanowires, particularly silicon nanowires, have garnered significant scientific attention in the last 25 years and are considered a highly promising material for nanoscale devices and integrated circuits. Crucial factors for realizing applications through a bottom-up approach include the precise control of chemical composition, structure, size, morphology, and doping in semiconductor nanowire systems. In the realm of semiconductor nanowires, silicon nanowires stand out due to their distinctive physical and chemical properties, holding potential for a diverse range of applications, including field-effect transistors (FETs). However, the development of such devices based on individual silicon nanowires encounters challenges related to intricate fabrication processes and the limited reproducibility of their electrical characteristics. These challenges stem from various issues such as variations in nanowire length, diameter, and doping, as well as concerns with wire surface passivation, insufficiently controlled gate length, and silicidation complications arising from the annealing step (Cui et al., 2003).

The advancement of carbon-based nanomaterials has significantly facilitated breakthroughs in various fields. Recently, carbon nanotube-based nanocomposites (CNT-based nanocomposites) have emerged as promising biomaterials for a broad spectrum of biomedical applications owing to their distinctive electronic, mechanical, and biological attributes. Numerous studies have highlighted the antimicrobial potential of nanocomposite materials like silver nanoparticles (AgNPs), polymers, biomolecules, enzymes, and peptides when combined with carbon nanotubes (CNTs). It is essential to comprehend the mechanisms governing the antimicrobial activity of these CNT-based nanocomposite materials, including the individual and synergistic effects on cells. This review consolidates information on the interaction between microorganisms and various types of CNT-based nanocomposites to elucidate their respective antimicrobial efficacy under different

conditions. Additionally, the review delves into the current developmental stage of CNT-based nanocomposite materials, addresses technical challenges, and explores the promising prospects of deploying these materials for their potential antimicrobial properties (Kang et al., 2008; Arjmand et al., 2022).

Palladium (Pd) nanocrystals, displaying diverse crystalline forms, showcase unique enzyme-like activities in the production of reactive oxygen species (ROS). The regulation of potential cytotoxicity by this crystallinity-dependent catalytic activity has yet to be fully understood. In this study, we synthesized Pd nanocrystals with four distinct crystalline forms and systematically unraveled the underlying mechanisms governing ROS-mediated cytotoxicity. Pd nanocrystals featuring {100} facets (nanocubes) and {111} facets (nanooctahedrons and nanotetrahedrons) induced cytotoxicity by generating singlet oxygen (102) and hydroxyl radicals (OH'), respectively. Simultaneously, Pd nanoconcavetetrahedrons, possessing both {110} and {111} facets, instigated ROS-mediated cytotoxicity through the activation of the superoxide (O2'-) pathway. The conversion of intracellular ROS led to the consumption of protons and the generation of hydroxide, resulting in pH alkalization and eventual cell death. The facet-dependent generation of ROS is highlighted as a crucial aspect of Pd nanocrystals. Furthermore, alkalization is identified as a novel biomarker for evaluating ROS-mediated cytotoxicity (Zhang et al., 2011).

Nanoparticles play a pivotal role in molecular imaging, a field that focuses on visualizing and understanding molecular processes within living organisms. These tiny particles, typically ranging from 1 to 100 nanometers in size, offer unique advantages in enhancing various imaging modalities and facilitating targeted imaging at the molecular level. Nanoparticles serve as contrast agents to enhance imaging signals. Their small size allows for efficient accumulation at the target site, improving the detection sensitivity of imaging techniques such as magnetic resonance imaging (MRI), computed tomography (CT), and ultrasound. Surface functionalization of nanoparticles enables specific targeting of molecular markers associated with diseases. Ligands or antibodies on the nanoparticle surface can selectively bind to biomolecules, guiding the nanoparticles to specific cells or tissues for more precise imaging. Nanoparticles can be designed to incorporate multiple imaging modalities, allowing for comprehensive and complementary information. For instance, a single nanoparticle may combine features suitable for MRI, fluorescence imaging, and positron emission tomography (Vega et al., 2023).

Some nanoparticles serve dual roles as both imaging agents and therapeutic delivery vehicles, a concept known as theranostics. These particles enable simultaneous imaging and treatment, offering a personalized and targeted approach to medicine. Nanoparticles can be engineered to be biocompatible, minimizing adverse effects on living tissues. Understanding their biodistribution is crucial for optimizing imaging efficacy and ensuring minimal toxicity. Nanoparticles, particularly those with fluorescent properties, are valuable for optical imaging techniques. Quantum dots and other nanomaterials with unique optical characteristics enable high-resolution imaging of molecular processes at the cellular and subcellular levels. Smart nanoparticles can respond to specific biological cues, such as changes in pH or the presence of certain molecules. This responsiveness enhances the specificity and sensitivity of imaging by providing real-time information about the biological environment. Surface modifications can extend the circulation time of

nanoparticles in the bloodstream, allowing for increased accumulation at the target site and prolonged imaging windows. Nanoparticles play a crucial role in advancing molecular imaging by offering versatile platforms for contrast enhancement, targeted imaging, multimodal capabilities, and theranostic applications. Their unique properties contribute to the development of more sensitive, specific, and personalized imaging approaches in biomedical research and clinical practice (Weissleder & Pittet, 2008).



Ongoing research in the field of nanoparticles

Highly water-soluble magnetite nanoparticles hold significant importance across various scientific and technological domains due to their unique properties and potential applications. Water-soluble magnetite nanoparticles can serve as contrast agents for MRI, providing enhanced imaging of biological tissues. Their water solubility is crucial for in vivo applications. The surface modification of magnetite nanoparticles enables the attachment of therapeutic agents, facilitating targeted drug delivery. Water solubility is essential for maintaining stability in physiological environments. Magnetite nanoparticles can be employed in water treatment processes for the removal of contaminants. Their water solubility ensures effective dispersion and interaction with pollutants. When exposed to an alternating magnetic field, water-soluble magnetite nanoparticles generate heat through hysteresis losses. This property is utilized in hyperthermia therapy for targeted destruction of cancer cells. Water-soluble magnetite nanoparticles are used as contrast agents in cellular imaging. Their ability to disperse in biological fluids without aggregation is crucial for accurate imaging and labeling of cells.

Magnetic Fluid Hyperthermia (MFH): The water solubility of magnetite nanoparticles is important for their use in magnetic fluid hyperthermia, where the nanoparticles are injected into tumors and heated using an external magnetic field to destroy cancer cells selectively. Highly water-soluble magnetite nanoparticles can be integrated into theranostic systems, providing both imaging and therapeutic functionalities. This is particularly valuable in personalized medicine. In catalysis, water-soluble magnetite nanoparticles can be employed due to their unique magnetic properties. These nanoparticles can facilitate reactions in

aqueous environments, making them suitable for various catalytic applications. Water solubility enhances the biocompatibility of magnetite nanoparticles, making them more suitable for biomedical applications and minimizing potential toxicity concerns. Water-soluble magnetite nanoparticles can be incorporated into biosensors for the detection of specific biomolecules. Their solubility ensures proper interaction with biological samples. The water solubility of these nanoparticles reduces the likelihood of aggregation, ensuring a uniform dispersion and improving their performance in various applications. In short, the significance of highly water-soluble magnetite nanoparticles lies in their versatile applications across medicine, environmental science, catalysis, and materials science. Their unique combination of magnetic properties and water solubility opens up opportunities for innovative solutions in a wide range of fields (Ge et al., 2007).

Addressing Challenges: The field of nanoparticles research faces several challenges that researchers and scientists are actively working to address. The potential toxicity of nanoparticles is a major concern, especially in biomedical applications. Understanding the long-term effects of exposure to nanoparticles on human health and the environment is crucial. The toxicity of nanoparticles is a complex and multifaceted area of research, and the effects can vary depending on factors such as nanoparticle composition, size, shape, surface charge, and exposure conditions. Numerous studies have investigated the potential toxicity of various nanoparticles, and the findings highlight the importance of understanding the biological interactions and potential risks associated with their use. It is important to note that the field of nanoparticle toxicity is dynamic, and ongoing research continues to expand our knowledge in this area (Neetika et al., 2023).

Standardizing methods for characterizing nanoparticles is indeed a challenging task due to the diverse nature of these materials, which can vary in terms of composition, size, shape, surface properties, and intended applications. However, efforts are being made to develop guidelines and standards to ensure consistency and comparability across different studies. Nanoparticles encompass a wide range of materials, including metals, metal oxides, polymers, and carbon-based materials. Standardizing methods for characterizing such diverse materials requires flexibility and adaptability. Nanoparticles can exist in various size ranges and shapes (e.g., spheres, rods, tubes), and these parameters significantly influence their properties. Developing standardized methods that accommodate this variability is challenging. Surface modifications play a crucial role in nanoparticle behavior and interactions. Different surface coatings and functionalizations can affect stability, toxicity, and other properties, making it challenging to create one-size-fits-all characterization methods. Various analytical techniques and instruments are used for nanoparticle characterization, each with its strengths and limitations. Standardization becomes challenging when different laboratories employ different instruments. The availability of well-characterized reference materials for nanoparticles is limited. Establishing a set of standard reference materials would aid in calibrating and validating measurement techniques. Nanoparticles can exhibit dynamic behaviors in different environments. Their properties may change over time due to agglomeration, degradation, or interactions with surrounding media. Collaborative efforts among researchers, institutions, and regulatory bodies on an international scale can help establish common standards and protocols for nanoparticle characterization. Creating well-characterized reference materials for various types of nanoparticles can serve as benchmarks for validation and calibration of